Unit 1 Study Guide: Introduction and Conversions	Name:
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Show your work. No work, no credit.

1) Solve the following problems using the factor-label (dimensional analysis) method:

*Be sure to memorize the metric prefixes for the test!

- a) How many mm are equivalent to 5.3256 m?
- b) How many km are equivalent to 4.56 m?
- c) How many µg are equivalent to 0.025 kg?
- d) If one atmosphere is equivalent to 760 mmHg, how many atmospheres are equal to 132 mmHg?
- e) If one mole of a substance has a mass of 18.01 g, how many moles of the substance are in 82.5 kg?
- 2) The literature value of the atomic mass of an isotope of nickel is 57.9 g/mol. If a laboratory experimenter determined the mass to be 59.6 g/mol, what is the percent error?
- 3) You fill a graduated cylinder to the 10.0 mL mark with water and then add 25.2 g of a metal. The water level covers the metal and the graduated cylinder reads 15.6 mL.
 - a) What is the density of the metal? What is its identity?
 - b) What volume would 0.120 kg of the metal occupy?
 - c) What mass will 75.3 cm 3 of the metal have? (1 cm 3 = 1 mL)

4) Fill in the missing information for each unit:

Unit	Name of unit	Fundamental (F) or Derived (D)	What it measures	Is this an SI unit?
m	meter		length of an object	
K				
J				
g				
mol				
cm^3				
οС				
S				
L				

5)	Put in scientific notation: a) 402,000:	b) 0.000301:
6)	Put in decimal (standard) form: a) 3.3 x 10 ⁻⁴ :	b) 6.1 x10 ⁷ :
7)	An adult Labrador retriever eats about 3 constants of 120 g, how many kg of food will the dog	ups of dry dog food per day. If each cup has a mass consume in a year?
8)	much time in days will the student invest or	s per day reviewing and working on chemistry, how ver an 85-day semester? (Bonus CHALLENGE: How s, hours, minutes, and seconds, instead of reporting
9)		ng. Joseph has measured the desks using his ruler 5 feet, 2.6 feet, and 2.5 feet. Are his measurements
10) Identify the variables - Does getting a good Independent: <u>Dependent:</u> <u>Controlled</u> :	
11)	
	mL 50 40 30 Close-up view	W 1000 ml — 600 — 400 — 200
Identi a)	fy the lab equipment. b) c)	d)
What a)	are appropriate uses for each piece of glassv	ware? b)
c)		