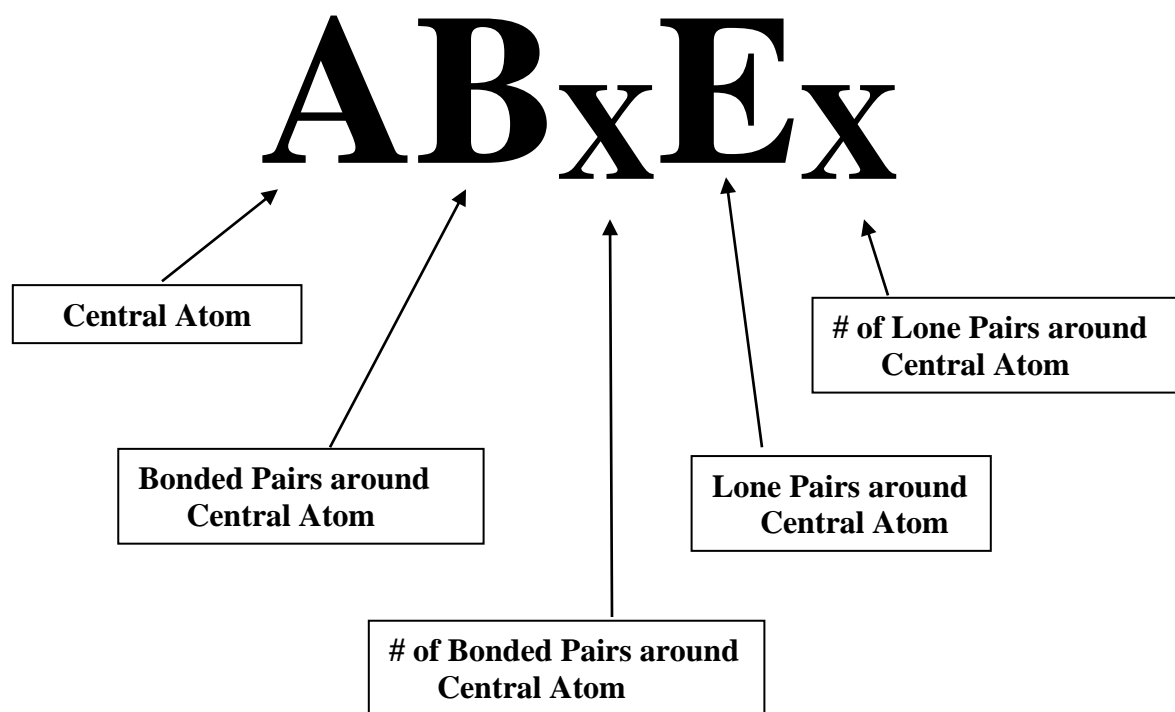


Molecular Geometry via A.B.E VSEPR Term



THINGS TO REMEMBER:

- Towards TOTAL Count: LONE pairs = ONE pair
- Towards TOTAL Count: Single, double, AND triple bonds around central atom = ONE pair
- Ex: EPG → *Trigonal Planar*
 - MG → *Bent*
 - VSEPR Term = AB_2E_1
- Ex: EPG → *Trigonal Bipyramidal*
 - MG → *T-shaped*
 - VSEPR Term = AB_3E_2