

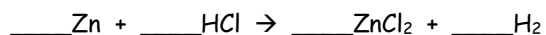
**EXTRA PRACTICE: Limiting and Excess Reactants Practice #2**

Name: \_\_\_\_\_

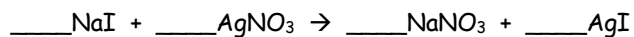
1. 50.0 grams of sodium metal reacts with 60.0 grams of chlorine to produce sodium chloride. What is the L.R. and E.R.?



2. In an S-R reaction, 50.0 grams of zinc reacts with 30.0 grams of hydrochloric acid (HCl) to produce zinc chloride. How many grams of zinc chloride are produced?



3. 40.0 grams of sodium iodide reacts with 50.0 grams of silver nitrate in a D-R reaction. How many grams of silver iodide is produced?



4. When 2.60 grams of hydrogen reacts with 10.2 grams of oxygen, water is produced. What is the L.R. and E.R.?



