

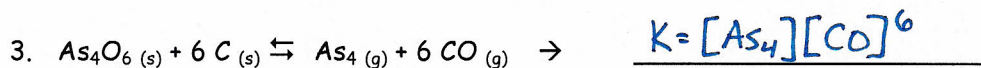
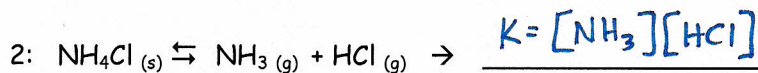
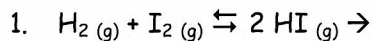
## ANSWER KEY

## EXTRA PRACTICE: Chemical Equilibrium Practice #2

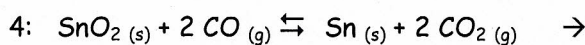
Name: \_\_\_\_\_

Write the equilibrium expression (K) for the following reactions:

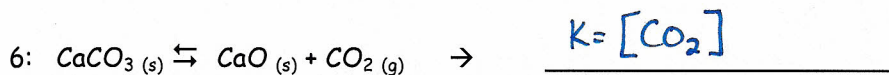
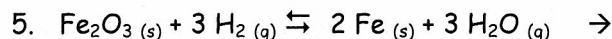
$$K = \frac{[\text{HI}]^2}{[\text{H}_2][\text{I}_2]}$$



$$K = \frac{[\text{CO}_2]^2}{[\text{CO}]^2}$$



$$K = \frac{[\text{H}_2\text{O}]^3}{[\text{H}_2]^3}$$



Balance each equation and then write the equilibrium expression (K) for each reaction:

