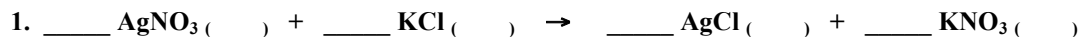


Net Ionic Equations Practice #2

Name: _____

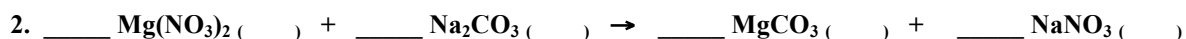
Complete the following net ion equations. Show ALL work for full credit. This includes **showing ALL CHARGES!** Balance ALL equations first. If no reaction occurs, you **MUST** still balance the equation, but write NR under net ionic equation. Carry through your **CHARGES** and **COEFFICIENTS** into the final net ionic equation. *Determine if each compound is aqueous (aq) or solid (s) first.*



Complete (Total) Ionic:

Spectator Ions:

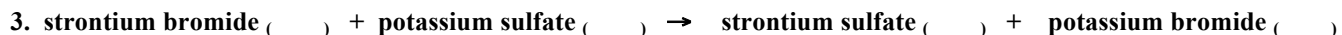
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

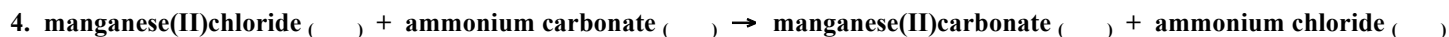
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

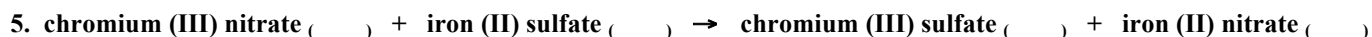
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:

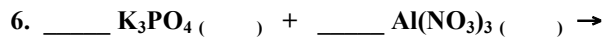


Complete (Total) Ionic:

Spectator Ions:

Net Ionic:

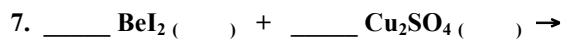
Predict the products, balance, and show the complete (total) ionic, spectator ions, and net ionic equations. Show all work!



Complete (Total) Ionic:

Spectator Ions:

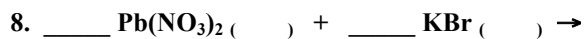
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

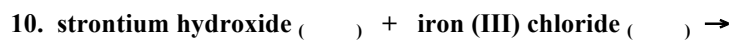
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic: