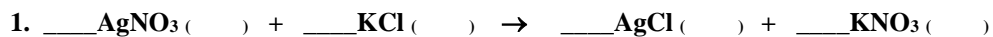


Net Ionic Equations Practice #1

Name: _____

Complete the following net ion equations. Be sure to show **ALL** of your work for full credit. Balance **ALL** equations first. If no reaction occurs, you must still balance the equation, but write **NR** under net ionic equation. Be sure to also carry through your charges and coefficients into the final net ionic equation. **Hint: Determine if each compound is aqueous (aq) or not first.**



Complete (Total) Ionic:

Spectator Ions:

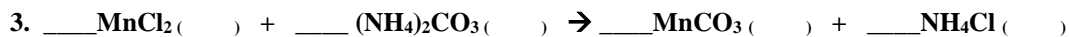
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

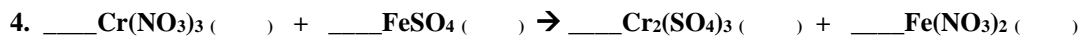
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

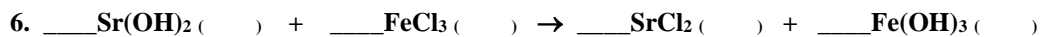
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic: