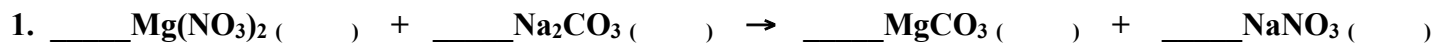


Net Ionic Equations Practice #1

Name: _____

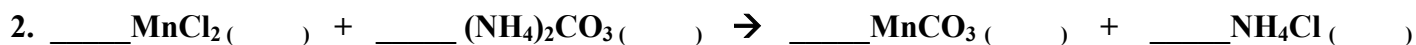
Complete the following net ion equations. Show ALL work for full credit. This includes showing ALL CHARGES! Balance ALL equations first. If no reaction occurs, you MUST still balance the equation, but write NR under net ionic equation. Carry through your CHARGES and COEFFICIENTS into the final net ionic equation. *Determine if each compound is aqueous (aq) or solid (s) first.*



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

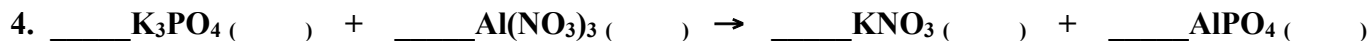
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

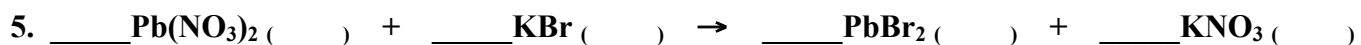
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

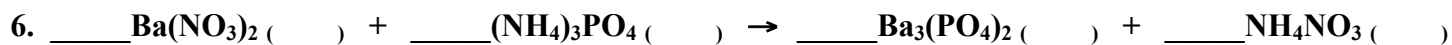
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

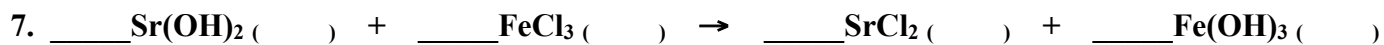
Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic:



Complete (Total) Ionic:

Spectator Ions:

Net Ionic: