## **Unit 4: Ch 8 – Nomenclature: Acids & Bases**

<u>ACIDS</u> : ≻	: <u>DEFINITION</u> –				
	o Howev	er, acids are form	ed by		<i>, <b>NOT</b></i> ionic.
>	FORMULA OF	ACIDS:			
	0				
	•	n =	→		compound
	•	X =ator	nic ORat	omic	<del></del>
	•	Method:			
>	NOMENCLATU	JRE: ACIDS			
	o 1	+	ANI	ON	
	•	Name: Use prefix			
	•				
		• Ex)	>	>	
		• Ex)	>	→	
	o <b>2</b>	+		<b>ANION</b> with	ending
	•		prefix		
	•	Name: Anion suff	ix ends in	_+	
	•	÷	·+	<del>-</del>	
		• Ex)	>	>	
		• Fv)	<b>→</b>	<b>-</b>	

o 3+		<i>'</i>	<b>4<i>NION</i></b> with	ending
•		prefix		
■ Name	e: Anion suffix ends i	n	_+	
•	>	+		
•	Ex)	$\rightarrow$	<b>→</b>	
•	Ex)	>	>	
SES:  DEFINITION –  NOMENCLATURE: B	ASES			
	л <u>ета</u> Иеtal		+	
o 2. Name:	+			
o 3. Method: _				
■ Ex)	>_		· >	
■ Ex)	>		. →	

BASES: