U1 – Ch 1-2 – Accuracy vs Precision/Significant Figures/Lab Measurements

Describing Measurements & Data:		
• QUALITATIVE –		
o Ex:		
• QUANTITATIVE –		
o Ex:		
Accuracy vs Precision:		
• ACCURACY –		
Accepted boiling point (BP) of water =		
• PRECISION –		
In Summary: Results of measurementsprecise measurement.	$_$ be more precise than the $_$	
Aim for accuracy precision!		
SIGNIFICANT FIGURES:		
• <u>DEFINITION</u> –		
 Precision of measurement is 	on instru	uments that reads
significant figures (_).
0	_ = More	measurement
SIGNIFICANT FIGURES RULES:		
o #1)		
o Evu	Evo	
○ Ex:	Ex:	

C	o +3)	Ex-A:		Ex	с-В:		
C	o +4)	Ex:		Ex	Œ		
C	o ====================================	Ex:		Ex	::		
C	° > #6)	Ex:		Ex	::		
	0	Ex:					
Readir	ng and I	Recording GRADUATE	D Measurements:				
0		lab instruments are		(ı	marked with		and
0	Record all values indicated by						
(uncertain) LAST digit ()	

o #2)